

Some Studies on the Uses of 2-Bromoethyl and 2-Iodoethyl Ester Blocking Groups in Peptide Synthesis: Samarium Diiodide-Mediated Deprotection

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Introduction

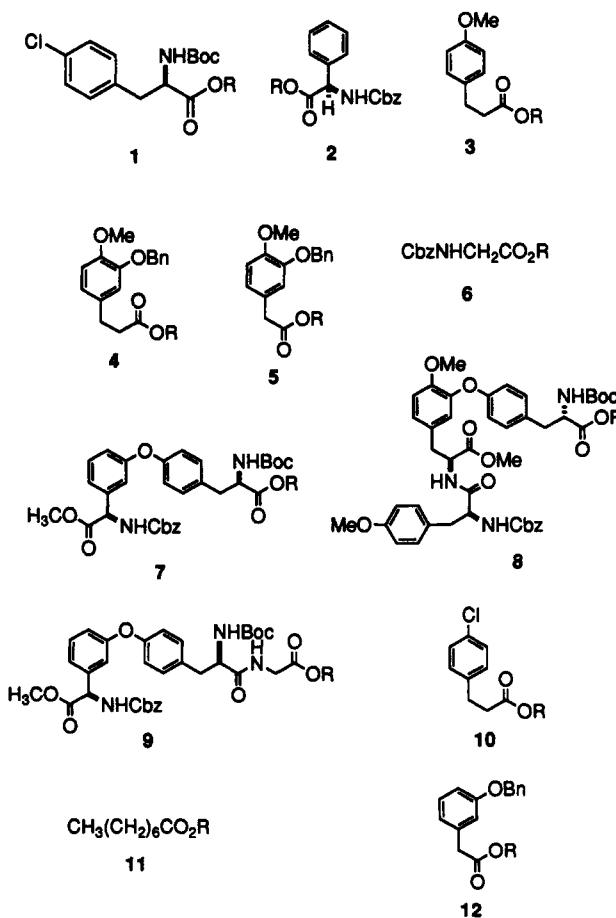
In connection with studies, currently underway in our laboratory, that are directed toward the synthesis of compounds related to the glycopeptide antibiotics ristocetin and vancomycin,¹ we required a carboxylate blocking group that is sufficiently robust to withstand a number of synthetic operations elsewhere in the molecule, yet can be removed under mild neutral conditions. The latter restriction is due to the fact that the target molecules contain a number of arylglycine residues that are prone to racemization under mildly basic conditions, and that *tert*-butoxycarbonyl and/or benzyloxycarbonyl groups were projected for amine protection during their assembly. A potentially useful candidate for our purposes is the 2-bromoethyl ester, which has been reported to be removable under a variety of nonbasic reaction conditions.² We were soon to learn, however, that this deprotection is highly problematic. In almost every case that we have studied, the 2-bromoethyl ester was converted to a 2-hydroxyethyl ester, probably via a classical neighboring group participation involving the carboxyl oxygen, which facilitates hydrolysis of the bromide.³ It has been reported that the 2-bromoethyl ester can be converted, via Finkelstein reaction, to a 2-iodoethyl ester.^{2c} The latter is somewhat easier to cleave reductively (Zn powder) and both two-step and one-pot procedures have been described for unmasking the carboxylate group.^{2c,4} In our hands, this tactic has also led to formation of 2-hydroxyethyl esters, either exclusively or as the major product.³ Magnus and co-workers have described similar difficulties during the deprotection of 2-chloroethyl carbamates, a problem that was solved by using samarium diiodide as the reducing agent.⁵ This paper reports on the use of SmI₂ for the deprotection of 2-iodoethyl esters under anhydrous conditions, which are compatible with the presence of arylglycines as well as a number of amine blocking groups, leading to the carboxylic acids in reproducibly excellent yields.

Results and Discussion

A number of 2-bromoethyl esters were prepared from the corresponding acids by following the standard method

using 2-bromoethanol/(DCC, pyridine) (Table 1). Direct deprotection of these esters by using SmI₂ was found to be extremely slow and inefficient, and so they were converted to 2-iodoethyl esters via Finkelstein reaction using NaI (5.0 equiv) in dry acetone for 7 h under nitrogen. This halide conversion was easily confirmed by ¹H and ¹³C NMR. Several 2-iodoethyl esters were also prepared directly from reaction of the carboxylic acid with 2-iodoethanol in the presence of DCC.

In the initial studies of deprotection by using SmI₂, *N*-Boc-*p*-chloro-*D*-phenylalanine 2-iodoethyl ester (**1b**) was used to determine the optimum reaction conditions. The reasons for choosing **1b** as model compound are that it has a relatively sensitive protecting group (*N*-Boc) and it is a chiral amino acid which is very closely related to intermediates that are currently being employed in our studies on vancomycin and ristocetin A. After reaction of **1b** with SmI₂ (7.0 equiv) at rt under deoxygenated, dry argon for 1 day, 75% was deprotected (¹H NMR); prolonged stirring (2 days) with excess of SmI₂ (10.0 equiv) did not lead to complete deprotection. At slightly higher temperature (35 °C, 7.0 equiv of SmI₂) the reaction was determined to be complete after 10 h from the ¹H NMR of crude product; at 40 °C, several products were observed, presumably due to instability of the *N*-Boc protecting group. For a comparison of *N*-Cbz and *N*-Boc, we investigated the deprotection of **2**, which showed that *N*-Cbz is stable toward SmI₂ at 40 °C. The optimized reaction conditions



For all structures: (a) R = CH₂CH₂Br; (b) R = CH₂CH₂I; (c) R = H

thus obtained (A: SmI₂ (6.0 equiv), 40 °C, 8 h; B: SmI₂ (7.0 equiv), 35 °C, 10 h) were used with several different

- (1) Pearson, A. J.; Park, J. G. *J. Org. Chem.* 1992, 57, 1744.
 (2) (a) Greene, T. W. *Protective Groups in Organic Synthesis*; John Wiley: New York, 1981. (b) Anderson, L. C.; Pinnick, H. W. *J. Org. Chem.* 1978, 43, 3417. (c) Kunz, H.; Buchholz, M. *Chem. Ber.* 1979, 112, 2145. (d) Ho, T. L. *Synthesis* 1975, 510. (e) Ho, T. L. *Synthesis* 1974, 715. For related methods, see Ho, T. L. *Synth. Commun.* 1978, 8, 301. (f) Scheffold, R.; Amble, E. *Angew. Chem., Int. Ed. Engl.* 1980, 19, 629. (g) Jacobson, R. M.; Clader, J. W. *Synth. Commun.* 1979, 9, 57. (h) Fried, J.; Sabo, T. *F. J. Am. Chem. Soc.* 1957, 79, 1130. (i) Joaquina, M.; Trigo, S. A. A.; Santon, M. I. A. O. *Collect. Czech. Chem. Commun.* 1988, 53, 2787. (j) Ho, T. L. *Synth. Commun.* 1978, 8, 359.
 (3) Park, J. G. Ph.D. Thesis, Case Western Reserve University, 1991.
 (4) Kunz, H.; Buchholz, M. *Liebigs Ann. Chem.* 1983, 11, 1859.
 (5) Magnus, P.; Anantharamayan, T. P.; Gallagher, T. *J. Chem. Soc., Chem. Commun.* 1982, 709.

Table 1. Yields for the Preparation and Deprotection of 2-Iodoethyl Esters^a

| entry | substrate | yield of b (%) | yield for b → c (%) |
|-------|-----------------|-----------------------|-----------------------------------|
| 1 | 1 ^b | 75 | 82 |
| 2 | 2 | 87 | 88 |
| 3 | 3 | 86 | 93 |
| 4 | 4 | 91 | 91 |
| 5 | 5 | 89 | 94 |
| 6 | 6 | 86 | 89 |
| 7 | 7 | 83 | 76 |
| 8 | 8 ^c | 95 | 70 |
| 9 | 9 ^d | 82 | 38 |
| 10 | 10 ^b | 80 | 90 |
| 11 | 11 ^b | 75 | 93 |
| 12 | 12 ^b | 73 | 88 |

^a Methods and spectroscopic data are given in the Experimental Section; for all structures: (a) R = CH₂CH₂Br; (b) R = CH₂CH₂I; (c) R = H. ^b These iodoethyl esters were prepared directly from the corresponding acid (RCO₂H/iodoethanol/DCC), as described for compound 1b. ^c The preparation of this compound is described in ref 8. ^d The preparations of 7a and 9a are described in refs 1 and 3.

compounds, and the results are summarized in Table 1. For simple 2-iodoethyl ester derivatives, reaction conditions A worked well and gave products of high purity in excellent yields. For amino acid derivatives, reaction conditions A (for *N*-Cbz) or B (for *N*-Boc) gave good to excellent yields. We next turned our attention to the possibility of racemization during the deprotection reaction. For this investigation, the racemization-prone phenylglycine derivative 2 was chosen as a model compound. Conversion of 2c ($[\alpha]_D^{25} = -121.2^\circ$, *c* 0.58, EtOAc) to 2a (2-bromoethanol, DCC, HOBT, DMF, CH₂Cl₂, 0 °C, 17 h, 86%) then to 2b (NaI, acetone, Δ), followed by deprotection of 2b, returned 2c without change in optical rotation ($[\alpha]_D^{25} = -121.0^\circ$, *c* 0.48, EtOAc), showing that no racemization occurs during the entire protection/halide exchange/deprotection sequence.

Conclusions

The 2-iodoethyl ester protecting group is removed successfully with SmI₂ under mild conditions. Theoretically, 2.0 equiv of SmI₂ is sufficient for this deprotection reaction, and the requirement for a large excess of this reagent may be due to its instability. In this context it may be noted that the deprotection does not proceed to completion unless oxygen is rigorously excluded, owing to competing oxidation of the samarium reagent. A two-step sequence (Finkelstein reaction, followed by deprotection with SmI₂) provides useful methodology for the deprotection of 2-bromoethyl esters. The latter protecting group is more stable than iodoethyl toward various reaction conditions, but its direct removal is problematic.

Experimental Section

SmI₂ (0.1 M in THF) was purchased from Aldrich Co. and used directly. All reactions using SmI₂ were performed under dry oxygen-free argon. All spectroscopic and general procedures were as described previously.¹

General Procedure for the Preparation of 2-Bromoethyl Esters and Direct Preparation of 2-Iodoethyl Esters. See the preparation of 6a and 1b, respectively.

General Procedure for Finkelstein Reaction. See the preparation of 9b.

General Procedure for the Deprotection of Iodoethyl Esters Using SmI₂. See the preparation of 2c and 7c.

***N*-[(1,1-Dimethylethoxy)carbonyl]-4-chloro-D-phenylalanine 2-Iodoethyl Ester (1b).** To a precooled (0 °C), stirred solution of *N*-Boc-4-chloro-D-phenylalanine (556 mg, 1.86 mmol)

in 20 mL of CH₂Cl₂ under N₂ were added 174 μL (1.2 equiv) of 2-iodoethanol followed by 301 μL of pyridine (2.0 equiv) by syringe. The mixture was stirred for 10 min at 0 °C and then 422 mg of DCC (1.1 equiv) was added in one portion. The resulting mixture was stirred for 16 h at 0 °C under N₂ in the dark. The reaction was quenched by the addition of 25 mg (0.15 equiv) of oxalic acid in 0.5 mL of THF and then it was allowed to come to rt and stirred for 30 min. The solid was filtered off and the filter cake was washed well with CH₂Cl₂ (10 mL × 3). The combined organic extracts were washed with 0.1 N HCl and brine and dried over MgSO₄, and the solvent was removed *in vacuo* to give a solid residue. Purification by flash chromatography on silica gel (Hex/EtOAc, 7/3) gave 629 mg (75%) of 1b as a white solid: mp 80.5–81.5 °C; *R*_f 0.37 (Hex/EtOAc, 7/3); IR (CHCl₃) 3438, 3026, 2982, 2934, 1744, 1711, 1493 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.28 (d, 2 H, *J* = 8.3 Hz), 7.11 (d, 2 H, *J* = 8.3 Hz), 4.96 (d, 1 H, *J* = 7.2 Hz), 4.59–4.57 (m, 1 H), 4.39–4.34 (m, 2 H), 3.28–3.00 (m, 4 H, CO₂CH₂CH₂I overlapping with ArCHHCH, ArCHHCH), 1.42 (s, 9 H); ¹³C NMR (75 MHz, CDCl₃) δ 171.1, 155.0, 134.4, 133.0, 130.7, 128.7, 80.2, 65.4, 54.2, 37.7, 28.3, -0.6; HRMS calcd for C₁₆H₂₁NO₂ClI 453.0206, found 453.0213.

***N*-[(1,1-Dimethylethoxy)carbonyl]-4-chloro-D-phenylalanine (1c):** yield 82%; mp 106.5–108.5 °C; $[\alpha]_D^{25} = -28.7^\circ$ (*c* 0.2, EtOAc); *R*_f 0.35 (EtOAc/MeOH, 9/1); IR (CHCl₃) 3438, 3026, 2982, 1719, 1653, 1493, 1221 cm⁻¹; ¹H NMR (300 MHz, DMSO-*d*₆) δ 12.62 (bs, 1 H), 7.32–7.24 (m, 4 H), 7.10 (d, 1 H, *J* = 8.8 Hz), 4.11–4.04 (m, 1 H), 3.00 (dd, 1 H, *J* = 13.7, 3.8 Hz), 2.80 (dd, 1 H, *J* = 13.7, 10.9 Hz); ¹³C NMR (75 MHz, DMSO-*d*₆) δ 173.4, 155.4, 137.1, 131.0, 128.0, 78.1, 55.0, 35.8, 28.1.

***N*-[(Phenylmethoxy)carbonyl]-D-phenylglycine 2-bromoethyl ester (2a):** yield 86%; mp 62–64 °C; $[\alpha]_D^{25} = -50.1^\circ$ (*c* 0.74, EtOAc); *R*_f 0.30 (Hex/EtOAc, 7/3); IR (CHCl₃) 3435, 3036, 2946, 1723, 1500, 1216 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.36–7.25 (m, 10 H), 5.78 (d, 1 H, *J* = 7.3 Hz, PhCH₂OCONHCH), 5.41 (d, 1 H, *J* = 7.3 Hz, PhCH₂OCONHCH), 5.13 (d, 1 H, *J* = 12.2 Hz, PhCHHOCO), 5.08 (d, 1 H, *J* = 12.2 Hz, PhCHHOCO), 4.40 (m, 2 H, CO₂CH₂CH₂Br), 3.41 (m, 2 H, CO₂CH₂CH₂Br); ¹³C NMR (75 MHz, CDCl₃) δ 170.4, 155.3, 136.0, 129.0, 128.7, 128.5, 128.2, 127.2, 67.2, 64.8, 58.0, 27.7; HRMS calcd for C₁₈H₁₈NO₄Br 391.0420, found 391.0408.

***N*-[(Phenylmethoxy)carbonyl]-D-phenylglycine 2-iodoethyl ester (2b):** yield 87%; mp 68.5–70.5 °C; $[\alpha]_D^{25} = -55.3^\circ$ (*c* 0.6, CHCl₃); *R*_f 0.36 (Hex/EtOAc, 7/3); IR (CHCl₃) 3435, 3014, 2967, 1742, 1723, 1602, 1517 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.37–7.26 (m, 10 H), 5.78 (d, 1 H, *J* = 7.1 Hz), 5.39 (d, 1 H, *J* = 7.1 Hz), 5.13 (d, 1 H, *J* = 12.1 Hz), 5.08 (d, 1 H, *J* = 12.1 Hz), 4.40–4.33 (m, 2 H), 3.49–3.17 (m, 2 H); ¹³C NMR (75 MHz, CDCl₃) δ 170.2, 155.3, 136.1, 136.0, 129.0, 128.7, 128.5, 128.2, 127.2, 67.2, 65.6, 58.0, -0.9; HRMS calcd for C₁₈H₁₈NO₄I 439.0282, found 439.0283.

***N*-[(Phenylmethoxy)carbonyl]-D-phenylglycine (2c).** To a stirred, precooled (0 °C) solution of 2-iodoethyl ester (2b, 10 mg, 0.18 mmol) in 10 mL of THF was added 10.9 mL (6.0 equiv, 0.1 M in THF) of SmI₂ by syringe. The resulting mixture was warmed to 40 °C and stirred for 8 h under deoxygenated argon. The mixture was cooled to rt, diluted with 30 mL of EtOAc, and quenched by the addition of 10 mL of 0.1 N HCl. The organic layer was separated and the aqueous layer was washed with EtOAc (10 mL × 2). The combined organic extracts were washed with 10 mL of 0.2 M sodium thiosulfate and brine and dried over MgSO₄, and the solvent was removed *in vacuo* to give a yellow residue. Purification by short column chromatography on silica gel (1 × 15 cm, EtOAc/MeOH, 95/5) afforded 46 mg (88%) of 2c as a white solid: mp 127.5–129.5 °C; $[\alpha]_D^{25} = -121.0^\circ$ (*c* 0.48, EtOAc); *R*_f 0.21 (EtOAc/MeOH, 9/1); IR (CHCl₃) 3436, 3026, 1725, 1667, 1497 cm⁻¹; ¹H NMR (300 MHz, DMSO-*d*₆) δ 12.86 (bs, 1 H, PhCHCO₂H), 8.11 (d, 1 H, *J* = 8.1 Hz, PhCH₂OCONH), 7.41–7.30 (m, 10 H), 5.16 (d, 1 H, *J* = 8.1 Hz, PhCHCO₂H), 5.04 (s, 2 H, PhCH₂OCONH); ¹³C NMR (75 MHz, DMSO-*d*₆) δ 172.0, 155.8, 137.1, 136.9, 128.4, 127.7, 65.6, 58.0; HRMS calcd for C₁₆H₁₅NO₄ 285.1001, found 285.1001.

3-(4-Methoxyphenyl)propionic acid 2-bromoethyl ester (3a): yield 72%; *R*_f 0.56 (Hex/EtOAc, 6/4); IR (CHCl₃) 3014, 1734, 1514 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.11 (d, 2 H, *J* = 8.6 Hz), 6.82 (d, 2 H, *J* = 8.6 Hz), 4.35 (t, 2 H, *J* = 6.1 Hz, CO₂CH₂CH₂Br), 3.76 (s, 3 H, OCH₃), 3.45 (t, 2 H, *J* = 6.1 Hz, OCH₂CH₂-

